

EFFECT OF COENZYME Q₁₀ ON ASCORBIC ACID, VITAMIN E, AND COENZYME Q CONCENTRATIONS IN TESTES OF RATS CHRONICALLY EXPOSED TO CADMIUM. S. Z. Pavlović¹, Branka I. Ognjanović², A. Š. Štajn², R. V. Žikić² and Zorica S. Saičić¹. ¹Department of Physiology, Siniša Stanković Institute for Biological Research, 11060 Belgrade, Serbia; and ²Institute of Biology, Faculty of Science, University of Kragujevac, 34000 Kragujevac, Serbia.

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Cadmium (Cd) has been recognized as being among the most toxic environmental and industrial pollutants, one that is present in soils, sediments, air, and water (Stoeppler, 1991). After penetrating into the organism, mostly through the respiratory and gastrointestinal tracts, Cd accumulates in the liver and kidneys, as well as in other tissues and organs, causing many metabolic, histological, and pathological changes, such as nephrotoxicity, cardiotoxicity, increased lipid peroxidation and hemorrhagic lesions of seminal tubules. In tissues, Cd also binds to proteins of low molecular mass, producing metallothioneins by the induction of metallothionein mRNA synthesis (George *et al.*, 1996). Cadmium depletes glutathione and protein-bound sulfhydryl groups, resulting in enhanced production of reactive oxygen species (ROS). These ROS result in increased lipid peroxidation, enhanced excretion of urinary lipid metabolites, modulation of intracellular oxidized states, DNA damage, membrane damage, altered gene expression, and apoptosis (Kim *et al.*, 2003).

The main antioxidant functions of coenzyme Q₁₀ (CoQ₁₀) are: to inhibit the process of lipid peroxidation and regenerate the active form of vitamin E (Ernst and Beyer, 1991); and to stabilize extracellular ascorbate in the organism (Gomez-Diaz *et al.*, 1997). It is known that CoQ₁₀ protects DNA from oxidation caused by lipid peroxidation and also protects the organism from oxidative stress induced by various toxic agents (Ernst and Dalner, 1995).

Our experiments were carried out with male 60-day-old white rats of Wistar strain weighing 190 ± 20 g at the onset of experiments. They were kept in individual cages under controlled conditions (light on from 5 a.m. to 5 p.m.; temperature of 23 ± 2°C) and had free access to water and food. The animals were divided into four experimental groups and treated for a period of 30 days. The first group of animals was the control (C, drinking tap water). The second group was treated with cadmium (Cd, 200 mg of CdCl₂ × 5H₂O/l of drinking water for 30 days + 100 µl of olive oil, i.m., every fifth day). The third group was treated with coenzyme Q₁₀ (CoQ₁₀, 40 mg of CoQ₁₀/ml dissolved in olive oil, i.m., every fifth day, drinking tap water). The fourth group was treated with cadmium and coenzyme Q₁₀ concomitantly (Cd+CoQ₁₀ in the above mentioned amounts). The

average intake of 17 mg Cd/day/kg body mass was calculated from water consumed during the 30-day treatment. The average intake of CoQ₁₀ was 16 mg/kg body mass every fifth day. All animals were invariably decapitated between 8 and 10 a.m. to avoid any possible rhythmic variations in the antioxidant level. The testes were dissected out within 3 min and prepared for further analysis. All chemicals were from Sigma (St. Louis, MO, USA). The concentration of ascorbic acid (AsA) was determined spectrophotometrically by the dinitrophenylhydrazine method (Roche, 1957). Vitamin E (Vit E) concentration was assayed by the method of Desai (1984) using bathophenanthroline. Coenzyme Q (CoQ) was determined by the method of Beyer (1989). Protein content was determined by the method of Lowry *et al.* (1951) using bovine serum albumin as a reference. Statistical analysis of the results was based on the Student's t test considering the a level of p < 0.05 to be significant (Hoeft, 1966).

Data on AsA, Vit E, and CoQ concentrations are given and compared to the control rats (C) in Table 1. The concentration of AsA was significantly increased in relation to the controls in all experimental groups, i.e., treated with Cd, with CoQ₁₀ concomitantly (p<0.005), and with Cd+CoQ₁₀ (p<0.01). Our earlier results on rat kidneys (Pavlović *et al.*, 2001) showed that Cd induces elevation of AsA concentration after chronic treatment. Increased concentration of AsA in testes of CoQ₁₀-treated animals tallies nicely with increased concentration of Vit E in these organs. It is known that AsA and Vit E are capable of acting synergistically as antioxidants and that each can exert sparing effects in relation to the other (Tanaka *et al.*, 1997). The data of Gomez-Diaz *et al.* (1997) also showed that CoQ₁₀ and its NADH-dependent reductase stabilize extracellular ascorbate in the organism. The concentration of Vit E was significantly higher in testes of animals treated with Cd, with CoQ₁₀, and with both Cd and CoQ₁₀ concomitantly than in testes of the controls (p<0.005). Vitamin E radical (Vit E•, α-tocopheroxyl radical) formed in the reaction with free lipid radicals (LOO•) would be regenerated by the reduced form of CoQ₁₀ (CoQ₁₀H₂), could explain the increased concentration of Vit E in the testes of rats treated with CoQ₁₀ (Ernst and Forman - Andrade, 1993). Increased concentrations of Vit E

Table 1. Concentrations of ascorbic acid (AsA), vitamin E (Vit E), and coenzyme Q (CoQ) in the testes of control rats (C) and rats treated with cadmium (Cd), with coenzyme Q₁₀ (CoQ₁₀), and with cadmium and coenzyme Q₁₀ concomitantly (Cd + CoQ₁₀). Values are means ± S.E. from seven animals.

TESTES	AsA (mg/100 g tissue)	Vit E (mg/g tissue)	CoQ (nM/mg proteins)
K	15.20 ± 0.97	7.76 ± 0.06	626.83 ± 15.81
Cd	22.30 ± 0.79 ****	12.39 ± 0.20 ****	609.41 ± 19.62
CoQ₁₀	32.42 ± 0.88 ****	12.86 ± 0.15 ****	593.37 ± 33.58
Cd + CoQ₁₀	19.03 ± 0.85 ***	14.61 ± 0.20 ****	622.30 ± 15.13

In relation to the controls (C): ***p<0.01; ****p<0.005 by Student's t test.

in the testes have a protective role against the toxic influence of Cd, which represents a physiological adaptation of the organism to toxic Cd effects. Contrary to AsA and Vit E, the concentration of CoQ in the testes was not significantly changed in any of the investigated groups of animals in comparison with the controls. Other authors performing dose- and time-dependent studies demonstrated that CoQ concentration after parenteral administration was increased only in the plasma and liver of rats (Scalori *et al.*, 1986).

It can be concluded that CoQ₁₀ administration to rats chronically exposed to exogenous Cd exerts beneficial effects on AsA and Vit E concentrations in the testes, resulting in improved antioxidant protection against Cd toxicity.

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References: Beyer, R.E. (1989). In: *CRC Handbook of Free Radicals*

and *Antioxidants in Biomedicine* (CRC Press), Boca Raton, 253-256. - Desai, I.D. (1984). *Methods Enzymol*, **105**, 138-147. - Ernster, L., and Beyer, R.E. (1991). In: *Biomedical and Clinical Aspects of Coenzyme Q* (Elsevier) Amsterdam, 45-58. - Ernster, L. and Dallner, G. (1995). *Biochem Biophys Acta*, **1271**, 195-204. - Ernster, L., and Fosmark-Andree, P. (1993). *Clin Investig*, **71**, S60-S65. - George, S.G., Todd, K., and Wright, J. (1996). *Comp Biochem Physiol*, **113C**, 109-115. - Gomez-Diaz, C., Rodriguez-Aguilera, J.C., Barroso, M.P., Villalba, J.M., Navarro, F., Crane, F.L., and Navas, P. (1997). *J Bioenerg Biomembr*, **29**, 251-257. - Hoel, P.G. (1966). In: *Introduction to Mathematical Statistics* (John Wiley and Sons), New York, 402-403. - Kim, S.C., Cho, M.K., and Kim, S.G. (2003). *Toxicol Lett*, **144**, 325-336. - Lowry, O.H., Rosebrough, N.J., Farr, A.L., and Randall, R.J. (1951). *J Biol Chem*, **193**, 265-267. - Pavlović, S.Z., Ognjanović, B.I., Štajn, A.Š., Žikić, R.V., Saičić, Z.S., and Petrović, V.M. (2001). *Arch Biol Sci*, Belgrade, **53**, 3P-4P. - Roe, H.J. (1957). In: *Methods of Biochemical Analysis* (Intersc. Publ.), New York, 115-139. - Scalori, V., Alessandri, M.G., Giovannini, L., and Bertelli, A. (1990). *Int J Tiss Reac*, **12**, 149-154. - Stoeppler, M. (1991). In: *Metals and their Compounds in the Environment* (VCH, Weinheim) New York, Basel, Cambridge, 803-851. - Tanaka, K., Hashimoto, T., Tokumara, S., Iguchi, H., and Kojo, S. (1997). *J Nutr*, **127**, 2060-2064.

